

Arylation of 3-Oxo-2,3-Dihydroindoles with Aryllead Triacetates

Jean-Yves Mérour^{a*}, Lucien Chichereau^a, Jean-Pierre Finet^{b*}.

^aLaboratoire de Chimie Bioorganique et Analytique associé au CNRS . Faculté des Sciences. BP 6759.
Université d'Orléans, 45067 Orléans Cedex 2, France.

^bLaboratoire Structure et Réactivité des Espèces Paramagnétiques associé au CNRS . Faculté des Sciences St Jérôme, 13397 Marseille Cedex 13, France.

Key Words : C-arylation ; substitutedphenylead triacetates : ketones : β -ketoesters : 2-arylindoles .

Abstract : Arylation of 2-alkoxycarbonyl-3-oxo-2,3-dihydroindoles with aryllead (IV) triacetates leads selectively to 2-alkoxycarbonyl-2-aryl-3-oxo-2,3-dihydroindoles. Direct selective C-arylation of 1-substituted-3-oxo-2,3-dihydroindoles with aryllead (IV) triacetates affords the corresponding 1-substituted-2-aryl-3-oxo-2,3-dihydroindoles.

2-Aryl-(2H)-indoles possess interesting antiinflammatory activity¹ or estrogen antagonist activity². The chemistry of 2-aryl-(2H)-indoles is well documented³ but with no or few functionnalities in the 3-position. 3-Oxo-2,3-dihydroindoles are key intermediates in the synthesis of biologically active compounds such as indomethacin⁴ or ellipticine⁵. The selective indirect mono C-2 alkylation⁶ and benzylation⁷ of 1-acetyl-3-oxo-2,3-dihydroindole were recently described. But introduction of an aryl substituent at the C-2 position is only possible by multistep indirect procedures, leading *inter alia* to compounds such as the stable free radicals, the 2,2-disubstituted-3-oxoindolin-1-oxyls⁸.

The efficiency of aryllead (IV) triacetates as regioselective arylating agents was discovered by Pinhey⁹. The regioselective arylation of heterocyclic systems^{10,11}, among them the C-2 arylation of 3-(2H)-benzofuranones¹¹, has recently been reported. Although direct α -arylation of ketones with lead derivatives has been reported in a limited number of cases^{9,11}, it generally requires activation of the ketone¹²⁻¹⁴. Up to now, the only reported result on the arylation of indolic compounds with aryllead reagents was the failure of indole to react with or without copper catalysis¹⁵. We now want to report our preliminary results on the arylation of 3-oxoindole derivatives.

The nitrogen-protected β -ketoesters **1a** ($R^1=H$) , **1b** ($R^1=Ac$) and **1c** ($R^1=Me$) were smoothly C-arylated by 4-methoxyphenylead triacetate **3**¹⁶ or 2,4-dimethoxyphenylead triacetate **4**¹⁶ ($CHCl_3$, pyridine, reflux, 5 hours) to the corresponding 2-aryl-2-alkoxycarbonyl-3-oxo-2,3-dihydroindoles **2**.

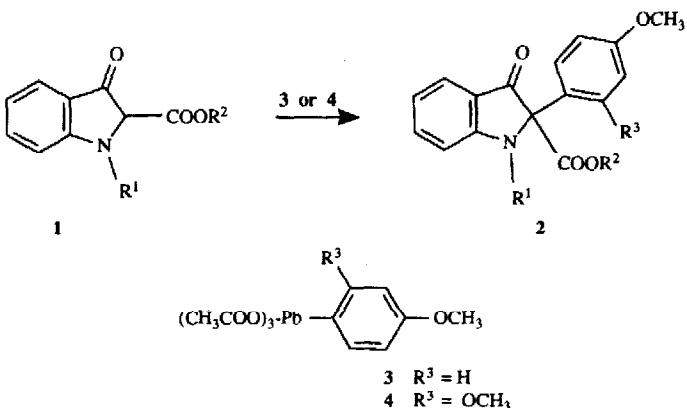


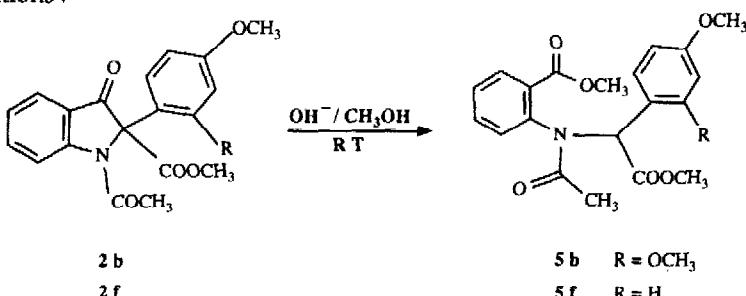
Table 1 : Arylation of 2-Alkoxy carbonyl-3-oxo-2,3-dihydroindoles 1 with Aryllead Triacetates 3 and 4 ^a

R ¹	R ²	1	R ³	Aryl lead	2 ¹⁷	Yield % ^b	mp °C
H	CH ₃	a	OCH ₃	4	a	68	184
COCH ₃	CH ₃	b	OCH ₃	4	b	62	135
CH ₃	C ₂ H ₅	c	OCH ₃	4	c	67	60
CH ₃	C ₂ H ₅	c	H	3	d	98	oil
H	CH ₃	a	H	3	e	95	141
COCH ₃	CH ₃	b	H	3	f	84	oil

^a Reactions were carried out in anhydrous chloroform at reflux; the molar ratio of 1 : 3 or 4 : pyridine was in all cases 1 : 1.1 : 3.3.

^b Yield data is for isolated homogeneous material.

Although the cleavage of the amide bond in the series of 1-acetyl-3-oxo-2,3-dihydroindoles is known to be easy ⁴, base-catalysed hydrolysis of the compound 2b and 2f did not lead to the expected 2a and 2e. Instead, compounds 5b and 5f were obtained, resulting from the five-membered ring opening by a retro-Dieckmann reaction on the sterically hindered β-keto-esters 2b and 2f, under very mild conditions .



In our studies on the arylation of benzofuranones, we have found that arylation of the unactivated ketone could be performed, albeit in poor yield and only with the very electron-rich 2,4,6-trimethoxyphenyllead triacetate¹¹. By contrast, the N-substituted-3-oxo-2,3-dihydroindoles **6** were easily arylated by less reactive reagents such as 2,4-dimethoxyphenyllead triacetate **4**. Indeed, a selective reaction was observed with monoarylation leading either to 2-monosubstituted 3-oxo-2,3-dihydroindoles (by arylation of 2-unsubstituted-3-oxo-2,3-dihydroindole) or to unsymmetrically 2,2-disubstituted-3-oxo-2,3-dihydroindole (by arylation of 2-monosubstituted 3-oxo-2,3-dihydroindole). Apart from the 5-methyl derivative **7f**, good yields were obtained in most cases and even a moderately good one was obtained for the synthesis of the bulky 2-phenyl-2-(2',4'-dimethoxyphenyl)-3-oxo-2,3-dihydroindole **7d**.

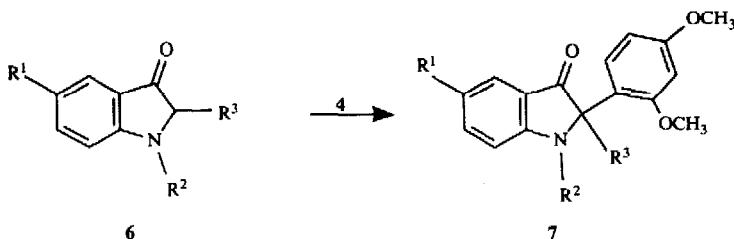


Table 2 : Arylation of 3-oxo-2,3-dihydroindoles **6** with 2,4-Dimethoxyphenyllead Triacetate **4**^a:

R ¹	R ²	R ³	6	7 ¹⁷	Yield % ^b	mp °C
H	COCH ₃	H	a	a	70	143
H	SO ₂ C ₆ H ₅	H	b	b	60	132
H	COCH ₃	CH ₃	c	c	72	147
H	COCH ₃	C ₆ H ₅	d	d	42	234
CH ₃ O	COCH ₃	H	e	e	81	188
CH ₃	COCH ₃	H	f	f	26	134

^{a, b} see Table 1.

Whereas in the corresponding oxygen-heterocycle benzofuranone the 2-aryl derivatives were only isolated from the palladium acetate-catalysed decarboxylation of the 2-allyloxycarbonyl derivatives, no such activation is required in the indole analogues as the unactivated ketone is directly and efficiently arylated. This might relate to a much higher percentage of enolic form in the 3-oxo-2,3-dihydroindole compared to the 3-oxo-2,3-dihydrobenzofuran series, enabling a facile formation of a covalent indole-3-oxylead intermediate.

Acknowledgements: We thank Prof. G. Guillaumet for helpful discussions.

References

- 1 Szmuszkovicz, J. ; Glenn, E.M. ; Heinzelman, R.V. ; Hester, J.B. ; Youngdale, G.A. *J. Med. Chem.*, 1966, 9, 527-536 .

- 2 a) Angerer, E.; Prekajac, J.; Strohmeier *J. Med. Chem.*, **1984**, *27*, 1439-1447; b) Angerer, E.; Knebel, N.; Kager, M.; Ganss, B.; *J. Med. Chem.*, **1990**, *33*, 2635-2640.
- 3 Sundberg, R.J. *The Chemistry of Indoles*; Academic Press: New York. 1970; pp 142-213.
- 4 Mérour, J.Y.; Coadou, J.Y.; Tatibouët, F. *Synthesis*, **1982**, 1053-1055.
- 5 Kilminster, K.N.; Sainsbury, M. *J. Chem. Soc., Perkin Trans 1*, **1972**, 2264-2267.
- 6 Buzas, A.; Mérour, J.Y. *Synthesis*, **1989**, 458-461.
- 7 Kawasaki, T.; Nonaka, Y.; Kobayashi, M.; Sakamoto, M. *J. Chem. Soc., Perkin Trans. I*, **1991**, 2445-2448.
- 8 a) Berti, C.; Colonna, M.; Greci, L.; Marchetti, L. *Tetrahedron*, **1977**, *33*, 2321-2327; b) Alberti, A.; Andruzzi, R.; Greci, L.; Stipa, P.; Marrosu, G.; Trazza, A.; Polini, M. *Tetrahedron*, **1984**, *44*, 1503-1515 and references therein.
- 9 For an excellent review on organolead triacetates, see: Pinhey, J.T. *Aust. J. Chem.*, **1991**, *44*, 1353-1382.
- 10 Barton, D.H.R.; Donnelly, D.M.X.; Finet, J.P.; Guiry, P.J. *Tetrahedron Lett.*, **1989**, *30*, 1539-1542; *ibid.*, **1990**, *31*, 7449-7452.
- 11 a) Barton, D.H.R.; Donnelly, D.M.X.; Finet, J.P.; Guiry, P.J.; Kiely, J.M. *Tetrahedron Lett.*, **1990**, *31*, 6637-6641; b) Donnelly, D.M.X.; Finet, J.P.; Kiely, J.M. *Tetrahedron Lett.*, **1991**, *32*, 3835-3836.
- 12 May, G.L.; Pinhey, J.T. *Aust. J. Chem.*, **1982**, *35*, 1859-1871.
- 13 Pinhey, J.T.; Rowe, B.A. *Aust. J. Chem.*, **1983**, *36*, 789-794.
- 14 Pinhey, J.T.; Rowe, B.A. *Aust. J. Chem.*, **1980**, *33*, 113-120.
- 15 Barton, D.H.R.; Yadav-Bhatnagar, N.; Finet, J.P.; Khamsi, J. *Tetrahedron Lett.*, **1987**, *27*, 3111-3114.
- 16 Willemsens, L.C.; De Vos, D.; Spierenburg, J.; Wolters J. *J. Organomet. Chem.*, **1972**, *39*, C61-62.
- 17 All new compounds gave spectroscopic data in agreement with the assigned structures;
compound 2b: IR(CHCl₃): 1750, 1725, 1685 cm⁻¹; ¹H-NMR(CDCl₃/TMS): δ=2.07(s, 3H, CH₃); 3.35(s, 3H, OCH₃); 3.81(s, 3H, OCH₃); 3.85(s, 3H, OCH₃); 6.36(d, J=2.4, 1H, H'₃arom); 6.55(dd, J=2.4, 8.7, 1H, H'₅arom); 7.23(t, J=8.5, 1H, Harom); 7.51(fd, J=7.7, 1H, H'₆arom); 7.66(fd, J=1.4, 7.4, 1H, Harom); 7.78(d, J=7.6, 1H, Harom); 8.40(br d, 1H, Harom); ¹³C-NMR(CDCl₃): 25.83(CH₃); 53.90, 55.41, 55.46(OCH₃); 77.99(CH); 99.50, 104.25, 116.54, 122.80, 123.91, 133.11, 136.91, 152.7, 158.30, 161.62(Carom); 165.96, 169.8(COO, NCO). 196.2(CO); MS(Cl) m/e 370(MH⁺);
compound 5f: yield 82%; mp 158°C; IR(CHCl₃): 1735, 1710, 1650 cm⁻¹; ¹H-NMR(CDCl₃): 1.82(s, 3H, CH₃); 3.65(s, 3H, OCH₃); 3.71(s, 3H, OCH₃); 3.74(s, 3H, OCH₃); 6.00(s, 1H, CH); 6.53(d, J=8.6, 2H, H'_{3,5}arom); 6.78(d, J=8.5, 2H, H'_{2,6}arom); 7.24(td, J=1.2, 6.8, H'₅arom); 7.48(td, J=2, 8, 1H, H₄arom); 7.62(dd, J=1.8, 8, 1H, H'₃arom); 7.76(dd, J=1.8, 8, 1H, H'₆arom); ¹³C-NMR(CDCl₃): 23.19(CH₃); 52.24, 52.49, 55.19(OCH₃); 63.23 (CH); 113.56, 124.40, 128.51, 130.19, 131.21, 131.41, 131.65, 132.97, 133.03, 139.79, 159.63, 165.42(Carom); 170.97, 171.46, 172.01(COO, NCO); MS(Cl) m/e 372(MH⁺); compound 7a: IR(CHCl₃): 1720, 1670 cm⁻¹; ¹H-NMR(CDCl₃): 2.05(s, 3H, CH₃); 3.79(s, 3H, OCH₃); 3.80(s, 3H, OCH₃); 5.59(s, 1H, CH); 6.42(dd, J=2.4, 8.7, 1H, H'₅arom); 6.51(d, J=2.4, 1H, H'₃arom); 6.88(d, J=8.7, 1H, H'₆arom); 7.25(fd, J=7.8, Harom); 7.69(fd, J=7.8, 1H, Harom); 7.75(fd, J=7.8, 1H, Harom); 8.6(fd large, 1H, Harom); MS(Cl) m/e 312(MH⁺).

Typical procedure: A mixture of 2-methoxycarbonyl-3-oxo-2,3-dihydroindole **1a** (0.535g, 2.8 mmol), (2,4-dimethoxyphenyl)lead triacetate **4¹⁶** (1.606g, 3.08mmol) and pyridine (0.728g) in anhydrous chloroform (30 ml) was refluxed for 5 hours; acidic treatment (HCl 1N), extraction with CH₂Cl₂ (2x20 ml), drying over MgSO₄ and evaporation in vacuo afforded a solid which was chromatographed over silica gel column; elution with CH₂Cl₂ gave **2a**; yield 0.671g(68%); mp 184°C; IR(CHCl₃): 3360(NH), 1725(COOCH₃), 1690(CO) cm⁻¹. ¹H-NMR (CDCl₃/TMS): 3.71(s, 3H, OCH₃); 3.76(s, 3H, OCH₃); 3.79(s, 3H, OCH₃); 6.44(dd, J=2.4, 8.5, H₅); 6.48(fd, J=2.4, 1H, H₃); 6.83(t, J=7.1, Harom); 6.88(d, J=8.2, Harom); 7.24(d, J=8.5, 1H, H₆); 7.45(t, J=7.7, Harom); 7.64(d, J=7.9, Harom); 5.80(s, 1H, NH); ¹³C-NMR (CDCl₃): 53.66, 55.54, 55.71 (OCH₃); 74.31(C₂); 99.5, 104.4, 113.06, 119.70, 125.27, 128.22, 138.02 (CH); 118.67, 119.79, 158.90, 161.06 (Cquat); 169.2(COOCH₃); 194.65(CO); MS (Cl) m/e 328(MH⁺).